- Superposition and Expectation values (example also covered in class)
- Free Particle
- Particle in a one-dimensional (1-D) box

Wavefunctions and Probability Density



 Think what might happen to the probability density when the quantum number n is very high

Particle in a box

 β-carotene is a linear polyene as shown below. It has 11 double bonds along a chain of 22 carbon atoms.



- Consider each C-C bong length to be 140 pm
- Question: Estimate the wavelength of light needed for excitation from ground state to the next excited state.





 $\Delta \boldsymbol{E} = \boldsymbol{hcl} \boldsymbol{\lambda}$

- Number of π e-s = 22
- $\Delta E = E_{12} E_{11}$

$$E_{12} = \frac{12^2 h^2}{8m_e L^2}$$
$$E_{11} = \frac{11^2 h^2}{8m_e L^2}$$
$$\Delta E = \frac{(12^2 - 11^2)h^2}{8m_e L^2}$$

- $m_{\rm e} = 9.109 \text{ x } 10^{-31} \text{ kg}$
- $h = 6.626 \times 10^{-34} \text{ J.s}$